Contents lists available at [ScienceDirect](http://www.sciencedirect.com/science/journal/00162361)

Fuel

journal homepage: www.elsevier.com/locate/fuel

Review article

A critical review of the $CO₂$ huff 'n' puff process for enhanced heavy oil recovery

^a Petroleum Systems Engineering, Faculty of Engineering and Applied Science, University of Regina, Regina, Saskatchewan S4S 0A2, Canada ^b State Key Laboratory of Oil and Gas Reservoir Geology and Exploitation, Southwest Petroleum University, Chengdu, Sichuan 610500, PR China

ARTICLE INFO

Keywords: $CO₂$ Enhanced oil recovery Heavy oil Huff 'n' puff Review study

ABSTRACT

Heavy oil resources have become increasingly important in recent years due to a reduction in light oil production and an increase in energy consumption. A large number of heavy oil reserves are found all over the world, and traditional production methods, such as solution gas drive, water flooding, etc., cannot gain a high heavy oil recovery factor, because of the high viscosity of the heavy oil. Although the thermal method has proven efficient and economical to produce heavy oil, it cannot be applied in deep reservoirs or reservoirs with thin pay zones due to the huge heat loss in these reservoirs. Thus, in order to enhance heavy oil production, several CO₂ injection processes are applied in heavy oil reservoirs. Among them, the $CO₂$ huff 'n' puff method has proven the most applicable.

In this research, the CO₂ huff 'n' puff process is reviewed in detail. Among the mechanisms of the CO₂ huff 'n' puff process in enhancing heavy oil production, the formation of foamy oil, viscosity reduction, and oil swelling are the most important ones, so that effect of foamy oil in the production stage is studied, and the viscosity reduction ratio with CO₂ injection and oil swelling factors at different temperatures and pressures are summarized. In addition, the diffusion coefficient, which indicates the mass transfer rate and amount of $CO₂$ dissolved into heavy oil through the two-phase interface of $CO₂$ and heavy oil, is analyzed in various heavy oil reservoirs at different temperatures and pressures.

Experimental studies on the $CO₂$ huff 'n' puff process indicate that the process applied in the heavy oil reservoir is successful and can be carried out with an oil viscosity up to 28,646 mPa·s and a reservoir permeability up to 24,200 mD. In pilot tests in the field, economical $CO₂$ huff 'n' puff processes have been applied in the heavy oil reservoirs with an oil gravity as low as 4 °API, a reservoir depth as high as 1985 m, and a pay zone as low as 12.2 m. Specifically, CO₂ utilization can be as low as 4.2 Mscf/Stb. Numerical simulation studies can gain very good simulation results on both experimental and pilot tests studies. However, mathematical models have seldom been published on CO₂ huff 'n' puff processes in heavy oil reservoirs.

1. Introduction

The heavy oil resource is defined as an asphaltic, dense, and viscous oil with an American Petroleum Institute (API) gravity less than 20 °API inclusive and a viscosity greater than 100 mPa·s [\[1](#page-9-0)-3]. Heavy oil resources are found all over the world, but they are mainly deposited in Canada and Venezuela. The total estimated volume of recoverable heavy oil (434 billion barrels) and bitumen (651 billion barrels) is almost the same as the remaining light oil reserves in the world. To meet the continuous increase in energy consumption, heavy oil production will be boosted in the future [\[1,4](#page-9-0)–9].

Two main difficulties for heavy oil production are high oil viscosity and thin oil pay zone. High oil viscosity leads to low mobility of the heavy oil in production process. To reduce high oil viscosity, two kinds of methods are mainly used:

- (1) Thermal methods reduce oil viscosity significantly due to the high temperature of the injected fluids. These methods include processes such as Steam Assisted Gravity Drainage (SAGD) [10-[12\],](#page-9-1) Cyclic Steam Stimulation (CSS) [\[13,14\],](#page-9-2) Steam Flooding [\[15,16\],](#page-9-3) in-situ combustion [17–[19\],](#page-9-4) etc.
- (2) Solvent based non-thermal methods reduce oil viscosity through the dilution of solvent into heavy oil. These methods include Cyclic Solvent Injection (CSI) [\[20,21\]](#page-9-5), Vapour Extraction (VAPEX) [\[22,23\]](#page-9-6), huff 'n' puff process [\[24,25\],](#page-9-7) etc.

<https://doi.org/10.1016/j.fuel.2017.11.092>

Received 11 August 2017; Received in revised form 7 November 2017; Accepted 22 November 2017 Available online 22 December 2017 0016-2361/ © 2017 Elsevier Ltd. All rights reserved.

[T](http://crossmark.crossref.org/dialog/?doi=10.1016/j.fuel.2017.11.092&domain=pdf)

[⁎] Corresponding author. E-mail address: fanhua.zeng@uregina.ca (F. Zeng).

Previous studies have determined that most heavy oil deposits in Canada are found in thin pay zones [\[26\]](#page-9-8). In Western Canada, up to 80% of the proven heavy oil reserves are in the oil pay zone, which is less than 5 m. Almost 97% of them are located in the pay zone, which is less than 30 m [27–[29\].](#page-9-9) The thermal based production method could not be implemented in this type of heavy oil reservoir due to the extremely high heat loss to the over-/under-burdens [\[26,30\]](#page-9-8). Regarding deep heavy oil reservoirs, thermal methods cannot enhance heavy oil production significantly, because the steam quality will be decreased remarkably when the steam is injected into the deep reservoir: leading to a very slow heat expansion in the reservoir. To avoid the negative effects of thin or deep reservoir properties on thermal based methods, the solvent-based non-thermal recovery method can be applied to enhance heavy oil recovery in the thin or deep heavy oil reservoir. Regarding the solvent based non-thermal method, methane [\[31,32\],](#page-9-10) ethane [\[33,34\]](#page-9-11), propane [\[22,35\]](#page-9-6), normal butane [\[22,36,37\]](#page-9-6), toluene [\[38,39\]](#page-9-12), $CO₂$ $[31,40,41]$, and mixture solvents $[20,24]$ etc., can be used as the injection solvent. Among the solvents, scholars have focused on $CO₂$ because (1) potential climate change will result in a rising temperature in the future and $CO₂$ emissions into the atmosphere (the latter, no doubt, is one of the key issues) [42–[48\]](#page-9-13); (2) the laboratory tests indicate that $CO₂$ can be absorbed in heavy oil and thus boost heavy oil production in the industry $[28,49]$; and (3) CO₂ can gain a much higher saturation pressure and higher viscosity reduction ratio at high pressure than other solvents [50–[52\]](#page-10-0).

There are many approaches to enhance heavy oil recovery by using the $CO₂$ injection process, including continuous $CO₂$ injection, intermittent CO_2 injection, water-alternate- CO_2 injection, and CO_2 huff 'n' puff. Among them, the $CO₂$ huff 'n' puff process is the most efficient process, although the recovery factor is still low [\[24,53\]](#page-9-7). An ongoing CO2 huff 'n' puff pilot test carried out in the Cold Heavy Oil Production with Sand (CHOPS) wells by Husky Energy in the Lloydminster area, Canada, indicates that the oil recovery has been increased about 8–20%, which is around 1.5 million barrels of heavy oil, and the recovery rate has been doubled $[54,55]$. Therefore, in this study, $CO₂$ huff 'n' puff process is reviewed to gain more details for future researches.

2. The $CO₂$ huff 'n' puff process

The $CO₂$ huff 'n' puff process is implemented in a single well. It is divided into three stages: (1) the injection stage, (2) the soaking stage,

and (3) the production stage [\[24,56](#page-9-7)–57], as shown in [Fig. 1](#page-2-0). In the injection stage, $CO₂$ is injected into the target formation through the well, which acts as the injector in the injection stage. The injected $CO₂$ bypasses the unmovable heavy oil and pushes part of the mobile heavy oil and water into a further location in the reservoir: leading to water saturation reduction near the wellbore so that the relative permeability of the heavy oil increases. The other part of the movable heavy oil is prevented from pushing away near the wellbore, and it is exposed in the injected $CO₂$ phase. The $CO₂$ diffusion process is negligible in the injection stage, because (1) the diffusion coefficient of $CO₂$ in heavy oil is not very high, (2) the injection stage is brief, and (3) the $CO₂$ is injected at a high injection rate. At the end of the injection stage, the pressure in the reservoir will be much higher than the reservoir pressure when the injection process started.

In the soaking stage, the well is shut-in. $CO₂$ diffusion occurs, and the key mechanisms of the $CO₂$ huff 'n' puff process in terms of oil swelling and viscosity reduction are obtained. During the soaking period, the mass transfer of $CO₂$ into heavy oil occurs and light/medium components in heavy oil are extracted into $CO₂$ so that the volume of heavy oil increases and the viscosity decreases.

In the production stage, when the well is open, part of the injected CO2, which does not dissolve into the heavy oil, is produced as the gas phase. Then the swelled heavy oil (indicated as a lighter color than the dead heavy oil in [Fig. 1\)](#page-2-0) that forms a large portion of the production fluids. Finally, heavy oil is produced with the water phase from a further location of the reservoir due to the drive force generated by the pressure drop. Part of the swelled oil is flushed by the movable water.

3. Mechanisms of the $CO₂$ huff 'n' puff in heavy oil reservoir

Regarding heavy oil reservoirs, the injected $CO₂$ is mainly under the immiscible condition for two reasons: (1) the Minimum Miscible Pressure (MMP) of the heavy oil is too high to be achieved in the heavy oil reservoir when the gravity of crude oil is lower than 30 °API [\[58\]](#page-10-2); and (2) the reduction of interfacial tension (IFT) between the injected $CO₂$ and heavy oil is not remarkable, so the miscible process cannot occur. The mechanisms of the immiscible $CO₂$ process are mainly reported as foamy oil, oil swelling, and viscosity reduction [\[51,56,59,60\]](#page-10-3), so they are discussed in detail. However, the upper aspects are insufficient to enhance heavy oil recovery in the $CO₂$ huff 'n' puff process alone [\[61\]](#page-10-4). The extra concepts are (1) reduction of interfacial tension,

Production stage

Fig. 1. Schematic of the $CO₂$ huff 'n' puff process in heavy oil reservoir. (a) oil drops are flushed to a further location in the reservoir by CO₂; (b) light/medium components are extracted by CO₂; (c) CO₂ dissolves into heavy oil; (d) oil drops connect together due to oil swelling; (e) the fluids flow direction; (f) residual oil due to wettability alteration; (g) oil drops generated from swelled oil in the soaking stage; and (h) oil drops in the water phase are driven from the further location.

(2) increased water wettability, (3) three-phase relative permeability effects, (4) $CO₂$ solubility in water, (5) light/medium hydrocarbons extraction, (6) solution gas drive, (7) $CO₂$ as a water shutoff agent, and (8) concomitant reactions with rock to increase the reservoir permeability near the wellbore, etc. [\[32,51,62](#page-9-15)–77].

3.1. Foamy oil

Compared with the $CO₂$ huff 'n' puff process in a heavy oil reservoir under water drive, the pressure depletion process that is conducted in the heavy oil reservoir can result in better oil production [\[78\]](#page-10-5) mainly due to foamy oil occurring in the production stage. Solution gas drive has been proven as the main production mechanism in the $CO₂$ huff 'n' puff process applied in a heavy oil reservoir [79–[82\].](#page-10-6) When $CO₂$ is injected into the heavy oil reservoir, it will dissolve into heavy oil through mass transfer, and the dissolved $CO₂$ gas will expand the volume of the heavy oil. Then the dissolved $CO₂$ will drive the heavy oil

out of the pores to the production well when the pressure is declined in the production stage. Because of the high viscosity of heavy oil, the phase of the $CO₂$ appears as gas bubbles, which are dispersed in heavy oil and flow with the heavy oil when the reservoir pressure declines. The produced heavy oil is a mixture containing small bubbles, this kind of produced fluid is defined as foamy oil [\[83](#page-10-7)–85]. The foamy oil phenomenon has been observed in experimental studies on the CO_2 huff 'n' puff process in heavy oil reservoirs, and it enhances heavy oil production significantly [\[60,76,86\]](#page-10-8).

In the production stage of the $CO₂$ huff 'n' puff process in heavy oil, the foamy oil phenomenon relates highly to the pressure decline rate, temperature, solvent solubility, etc. A higher pressure depletion rate results in a higher heavy oil recovery factor due to the higher pressure depletion rate producing more stable foamy oil in the production stage [\[7,87](#page-9-16)–89]. Considering the effect of temperature, researchers have found that the stability of the foamy oil decreases sharply and the volume of the dispersed gas increases with increasing temperature [\[90,91\],](#page-10-9) but an optimized temperature can be obtained for a certain oil sample [\[92\]](#page-10-10). Among different solvents, foamy oil, which is generated by using $CO₂$ saturated in heavy oil, can achieve a higher quality than other solvents (CH₄ or N_2) due to the slow desorption of CO₂ in heavy oil $[81]$. The solubility of $CO₂$ in heavy oil relates to injection pressure, as the CO2 solubility increases with the increasing of injection pressure. With higher $CO₂$ solubility, the foamy oil behavior will be more obvious in the production stage [\[76\],](#page-10-12) and the heavy oil recovery factor will be higher.

3.2. Viscosity reduction

Viscosity reduction is another main mechanism of the $CO₂$ huff 'n' puff process. Previous studies have indicated that the effect of viscosity reduction is more significant in heavy oil with a lower API gravity [\[24,93\].](#page-9-7) When $CO₂$ is recombined into heavy oil, the viscosity of the heavy oil is extremely reduced, as shown in [Table 1.](#page-3-0) The main reasons for viscosity reduction through $CO₂$ injection are: (1) the particulate matters in the heavy oil are washed out by the injected $CO₂$; (2) the viscous deposits are dissolved by the injected $CO₂$; (3) the viscous crude in heavy oil is diluted by the injected CO_2 ; and (4) the injected CO_2 is demulsified in heavy oil [\[94\].](#page-10-13) The viscosity reduction of heavy oil results in the fractional flow curve shifting to the right, so that the fractional flow of water is lower than that before $CO₂$ injection at the same water saturation and the oil mobility and oil connection are increased, which leads to a relative higher oil flow rate [\[56\].](#page-10-14)

The viscosity reduction ratio of a heavy oil- $CO₂$ system changes with the temperature, pressure, and solubility of the dissolved $CO₂$ [\[95\]](#page-10-15). [Fig. 2](#page-3-1) shows the viscosity reduction ratio and $CO₂$ solubility of a heavy oil -CO₂ system at different temperatures and pressures. With temperature increases, the viscosity of the dead heavy oil decreases extremely and the viscosity reduction ratios for the dead oil at 60 °C and 93 °C are 86.8% and 97.3%, respectively. Therefore, the effect of temperature on heavy oil viscosity is remarkable.

With $CO₂$ injection, when the temperature is lower than the critical temperature, the viscosity reduction mainly occurs at a lower pressure due to the mass transfer of the liquid phase being much slower than the gas phase. This leads to the effect of pressure on $CO₂$ solubility being not significant. Regarding the heavy oil- $CO₂$ system, the efficiency of viscosity reduction decreases with temperature increases at the same pressure due to (1) lower $CO₂$ solubility with higher temperature in heavy oil; and (2) lower viscosity of heavy oil at higher temperature, which results in less viscosity reduction potential. The experimental study indicates that a higher percentage of viscosity can be reduced by $CO₂$ injection for heavy oil with a higher viscosity [\[95\]](#page-10-15). With an increase in $CO₂$ solubility, the viscosity reduction ratio increases, which means a higher percentage of the heavy oil viscosity is reduced by injecting CO₂. The viscosity reduction ratio can be as high as 97% for the studied heavy oil.

|--|--|

Table 1 Summary of the measured viscosity reduction ratios and oil swelling factors in different heavy oils.

Fig. 2. The viscosity reduction ratio and $CO₂$ solubility of a heavy oil-CO₂ system at different pressures and temperatures [\[96\]](#page-10-19).

A summary of viscosity reduction studies on a heavy oil-CO₂ system is tabulated in [Table 1.](#page-3-0) [Table 1](#page-3-0) shows that the effect of $CO₂$ in heavy oil is significant, the viscosity reduction ratio can reach up to 99.8%, and the viscosity reduction of the heavy oil- $CO₂$ system relates to pressure and temperature. Among different heavy oil samples with a higher heavy oil viscosity, a greater viscosity reduction ratio can be obtained. Regarding the same heavy oil sample, the viscosity reduction ratio decreases with an increase in temperature.

3.3. Oil swelling

When $CO₂$ is injected into the heavy oil reservoir, an important phenomenon is observed in terms of oil swelling, because the injected $CO₂$ dissolves into the heavy oil and expands the volume $[71,101,108,109]$. The oil swelling is an important mechanism to enhance heavy oil recovery in the $CO₂$ huff 'n' puff process, because (1) oil swelling shows an advantage on the movable oil, and an inverse proportional relation is found between the oil swelling factor (which indicates the degree of oil swells and is defined as the volume of crude oil saturated with $CO₂$ at the reservoir pressure and temperature divided by the volume of crude oil at the atmospheric pressure and reservoir

temperature [\[110\]\)](#page-10-17) and the residual oil saturation; (2) the mobility of the heavy oil is improved; (3) the dissolved heavy oil will generate a drainage force to push water out of the pore space; and (4) oil swelling can increase the oil saturation, resulting in an increase of oil relative permeability, which increases the oil phase fractional flow in the production stage [\[32,33,58,72,111\].](#page-9-15)

The degree of oil swelling factor relates to pressure, temperature, and oil composition [\[58\]](#page-10-2). [Figs. 2 and 3](#page-3-1) indicate that the plots of the oil swelling factor have the same trends as the plots of $CO₂$ solubility, which means that, under the same conditions (temperature and pressure), the oil swelling factor is proportional to the $CO₂$ solubility. The effects of pressure on the oil swelling factor are different at different temperature, and a linear relationship is obtained between the oil swelling factor and pressure when the temperature is greater than the critical temperature. However, the phase of the $CO₂$ affects the oil swelling factor remarkably. In the low-pressure region (when the $CO₂$ is in the gas phase), the oil swelling factor increases with the pressure increases. At higher pressure, the phase of $CO₂$ changes from the gas phase to the liquid phase, leading to lower $CO₂$ solubility and a reduced effect of pressure on oil swelling factor. The effect of temperature shows

Fig. 3. Oil swelling factor of the heavy oil-CO₂ system at different pressures and temperatures [\[96\].](#page-10-19)

that a higher temperature leads to a lower oil swelling factor in the lowpressure region due to $CO₂$ solubility decreasing with temperature increases. In a higher-pressure region, the oil swelling is greater than that at low temperature due to the phase change reducing the $CO₂$ solubility. Regarding oil composition, lighter oil can get a higher oil swelling factor than that of heaver oil, because more $CO₂$ can be dissolved into the lighter oil [\[24\].](#page-9-7)

3.4. Diffusion coefficient

Another important parameter that impacts the properties of a heavy oil -CO₂ system is the diffusion coefficient, which indicates the diffusion rate and the amount of $CO₂$ dissolved into the heavy oil [\[112](#page-10-32)–118]. Previous studies indicate that heavy oil production in the vapour-extraction process is mainly from the transient zone, where heavy oil is saturated with an injected solvent and the area of the transient zone is controlled by the molecular diffusion rate of the injected solvent [\[119,120\].](#page-10-33) As a type of solvent, when $CO₂$ is injected into the heavy oil reservoir, it is gradually dissolved into the heavy oil by means of mass transfer (molecular diffusion), especially in the soaking stage [\[111,121\].](#page-10-34) This results in viscosity reduction and oil swelling so that heavy oil production can be enhanced.

The diffusion coefficient relates to pressure, temperature, and oil composition, as shown in [Fig. 4](#page-4-0) and [Table 2.](#page-5-0) The effect of pressure on the diffusion coefficient is more sensitive at a higher temperature than at a lower temperature because (1) the lower surface tension of oil molecules can be obtained at a higher temperature, so that the mass transfer rate of $CO₂$ molecules into heavy oil is higher and (2) a lower heavy oil viscosity is obtained at a higher temperature and $CO₂$ molecules can pass through the interface easier. Kavousi et al. studied the CO2 diffusion coefficient in heavy oil at different temperatures and pressures $[122]$. In their experimental researches, the $CO₂$ diffusion coefficient increases with pressure increases. However, if the pressure continues increasing at a very high level, the viscosity and density of the heavy oil- $CO₂$ system increases as well, which causes the diffusion coefficient decreases steadily [\[123\]](#page-11-1).

Under a constant temperature, the diffusion coefficient increases with pressure increases in the relative lower pressure region, mainly because the higher pressure supports a greater drive force for the $CO₂$ transferring into heavy oil. The combined effects of pressure and temperature show that the diffusion coefficient of $CO₂$ in heavy oil increases with pressure and temperature increase.

The viscosity of heavy oil decreases with temperature increases, which can be concluded as the diffusion coefficient of $CO₂$ in heavy oil decreasing with heavy oil viscosity increases. [Table 2](#page-5-0) indicates that $CO₂$ diffusion coefficients are different for different oil samples. Even different diffusion coefficients can be achieved using the same experimental results [\[124,125\],](#page-11-2) due to (1) different treatment of the pseudocomponents for the heavy oil in the calculation, and (2) a slight difference between the objective functions.

Regarding $CO₂$ solubility in heavy oil, the solubility of $CO₂$ increases with pressure increases and decreases with temperature increases, but no significant relationship can be found with the $CO₂$ diffusion coefficient.

To measure the diffusion coefficient of $CO₂$ in heavy oil, direct and indirect measurement methods have been applied in previous studies. In the direct method [\[126,127\]](#page-11-3), oil samples are extracted out of the tested system during the test to involve compositional analysis. Then a mathematical model is required to calculate the diffusion coefficient. Experimental errors in the direct method are not easily avoided. In the indirect method, the properties of heavy oil and $CO₂$ are measured and the changes are monitored during the tests. The tested properties include pressure decay monitoring [\[32,112,126,127\],](#page-9-15) volume changing measurement [\[128\]](#page-11-4), volatilization rate of solvent testing [\[129\]](#page-11-5), location of the gas-liquid interface determination [\[112,130\],](#page-10-32) etc. Other indirect methods such as dynamic pendant drop volume analysis [\[33\]](#page-9-11),

Nuclear Magnetic Resonance (NMR) [\[131\],](#page-11-6) and X-ray Computer Assisted Tomography (CAT) [\[132\]](#page-11-7) are also used to determine the concentration of $CO₂$ at different locations of the test fluids.

The diffusion coefficients measured by previous scholars are summarized in [Table 2](#page-5-0). [Table 2](#page-5-0) indicates that the diffusion coefficient of $CO₂$ in heavy oil relates to oil components, viscosity, temperature, and pressure, and that most of the measured diffusion coefficients of $CO₂$ in heavy oil are in the order range 10^{-10} m²/s- 10^{-9} m²/s. For heavy oil with a higher API gravity, there are more light or medium components, which results in the $CO₂$ diffusion process occurring easily, so that the $CO₂$ diffusion coefficient is higher than that in heavy oil with a lower API gravity. Compared with different heavy oil samples, the heavy oil sample with a relative lower viscosity is beneficial to the diffusion coefficient, so that a greater diffusion coefficient can be obtained. Regarding the test methods, pressure decay is the most popular method in previous studies and the pressure profile is matched using the derived mathematical models. Then the diffusion coefficient of $CO₂$ is calculated. In the calculation process, the diffusion coefficient of $CO₂$ differs slightly according to different boundary conditions (equilibrium, quasiequilibrium, and non-equilibrium) in the mathematic model even through the same tests are applied.

4. Experimental studies

Prior to the implementation of pilot tests of the $CO₂$ huff 'n' puff process in the heavy oil field, the applicability study on this process was carried out in the laboratory. The properties of the heavy oil- $CO₂$ system, in terms of viscosity, $CO₂$ solubility, swelling factor, etc., were studied in detail to investigate fluid properties in the process. The reservoir properties including porosity, permeability, oil saturation, and water saturation were researched to determine the reservoir properties that would be suitable for the $CO₂$ huff 'n' puff process. In addition, the operation parameters, such as number of cycles, injection pressure, soaking time, pressure depletion rate, etc., were optimized at the lab scale.

[Table 3](#page-6-0) indicates that the successful application of the $CO₂$ huff 'n' puff process at the lab scale can be carried out under high reservoir temperature as high as 90 °C, with an extra-high heavy oil viscosity that reaches to 28,646 mPa·s. The permeability of the tested models ranges from 30 mD, which is real core, to 24,200 mD, which is packed by using sands. The recovery factor did not change too much, so that the permeability is not a sensitive parameter that affects the production performance of the $CO₂$ huff 'n' puff process applied in a heavy oil reservoir. The oil saturation in the physical models shows that the process can be applied even in a low oil saturation reservoir, which can be as low as 40.6%, which means that this process can be conducted in the reservoir under higher water saturation [\[76\].](#page-10-12)

Fig. 4. $CO₂$ solubility and diffusion coefficient at different temperatures and pressures [\[122\].](#page-11-0)

Table 2

Summary of the measured diffusion coefficients of $CO₂$ in different heavy oils.

Also, the application of the $CO₂$ huff 'n' puff process in a heavy oil reservoir with an extremely high water cut (98%) was studied by using real core plugs [\[76,141\]](#page-10-12). The experimental results show that the $CO₂$ huff 'n' puff process can be applied in the heavy oil reservoir with a high water cut. Regarding the operation parameters, to optimize the number of cycles, a maximum of 10 cycles were conducted in the lab test, and it was found that the highest production rate occurred in the second cycle [\[68,82\].](#page-10-35)

The injection pressure ranges from 1724 kPa to 25,000 kPa, which relates to the reservoir permeability. The soaking time is mainly around 2 days, but it can be as high as 18 days. In the production stage, the pressure depletion rate affects the heavy oil reservoir significantly. The trend shows that oil recovery increases with increasing pressure depletion rate.

The properties of heavy oil saturated with $CO₂$ were studied by using three different heavy oil samples with oil gravities of 10, 15, and 17 °API, respectively [\[69,96\].](#page-10-36) The experimental results indicate that, without $CO₂$ solution, the viscosity of the oil sample increases with pressure increases at a constant temperature. With $CO₂$ recombination, the $CO₂$ solubilities of the three heavy oil samples increase with increasing pressure at different temperatures. However, the trends of heavy oil properties change at different temperatures when $CO₂$ is recombined into heavy oil. At a lower temperature, $CO₂$ solubility increased until the critical pressure was reached, and then it kept almost stable. The same phenomena were observed for heavy oil viscosity and swelling factor. At higher temperatures, with increasing pressure, (1) $CO₂$ solubility in heavy oil increases, (2) the viscosities of the heavy oil-CO2 system decreases significantly, and (3) the swelling factor increases up to as high as 1.28 of the tested oil samples.

The operation parameters, including injection pressure and soaking time, were researched by using a physical Berea core [\[68,82\]](#page-10-35). The experimental results indicate that, (1) the maximum oil production rate occurred at the second cycle; (2) higher oil recovery factor can be obtained when higher operation pressure was conducted, due to more $CO₂$ could be dissolved in the heavy oil to reduce the viscosity and increase the oil swelling factor at higher pressure; (3) even through the longer

soaking time could not improve the oil recover factor remarkably, the oil production rate in the first several cycles could be increased significantly. Another important application of $CO₂$ huff 'n' puff process is conducted to enhance heavy oil production after solution gas drive process [\[25,31,41\].](#page-9-17) The results of these studies show that, higher recovery factor was obtained in the first cycle when longer primary production duration was carried out; the effect of surfactant in the test did not remarkable; and more than 30% of the total recovery factor can be improved because of the $CO₂$ huff 'n' puff process.

To obtain a better understanding, the same process was carried out on two kinds of heavy oil samples in Shi and Kantzas's research [\[31\]](#page-9-10). The effects of oil viscosity and pressure depletion rates were analyzed respectively. The tests indicate a high potential for heavy oil production using the $CO₂$ huff 'n' puff process, even in the low residual oil saturation (47.1% OOIP) after the primary production test.

In order to study the mechanisms of the $CO₂$ huff 'n' puff process applied after primary production in the heavy oil reservoir, Lu et al. conducted experiments in both a micromodel and a sandpack [\[25\]](#page-9-17). The micromodel tests show that the conversion pressure (at which the primary production process was changed into the huff 'n' puff process) affects oil production significantly, and the pseudo-bubble point pressure is the optimized conversion pressure. The results of the sandpack tests show that the highest recovery factor was obtained when the conversion pressure was applied, and the heavy oil recovery factor of the $CO₂$ huff 'n' puff process is proportional to injection pressure, soaking time, and pressure depletion rate. Therefore, CO₂ huff 'n' puff can be applied in the heavy oil reservoir after the primary production process due to the high potential of $CO₂$ huff 'n' puff as a follow-up process to primary recovery.

5. Pilot test studies

Pilot tests of the $CO₂$ huff 'n' puff process in the heavy oil reservoirs were carried out in several oil fields in the past several decades. Most of the published cases are successful. To analyze the production performance of heavy oil under the $CO₂$ huff 'n' puff process, the properties of

Table 3

Heavy oil recovery factor of the remaining oil in place (ROIP). Heavy oil recovery factor of the remaining oil in place (ROIP).

Oil saturation after primary production process Oil saturation after primary production process.

3

4

Irreducible-water saturation.

Irreducible-water saturation.

the reservoirs and the production performance of the process are

summarized in [Tables 4 and 5](#page-7-0), respectively. [Table 4](#page-7-0) indicates that the properties of the heavy oil reservoirs in which successful CO₂ huff 'n' puff processes were carried out affect the production performance to some extent. The comparison of the heavy oil viscosity indicates that the successful application of the CO_2 huff 'n' pu ff process in a heavy oil reservoir can be carried out not only in a medium heavy oil reservoir with a viscosity as low as 118 mPa·s, but also in an extra-heavy oil reservoir with a viscosity as high as 17,200 mPa·s. In the studied reservoirs, the temperature, which affects the viscosity of heavy oil signi ficantly, ranges from 33.3 °C to 65.5 °C. In this temperature range, di fferent CO ² phases (vapour phase and supercritical phase) were obtained in the pilot tests of the $CO₂$ huff 'n' puff process in the heavy oil reservoir. Because the reservoir temperatures in the pilot tests are higher than the critical temperature of CO_2 (31.1 °C), the phase of the injected $CO₂$ only relates to the initial reservoir pressure. As listed in [Table 4](#page-7-0), in most of the tested reservoirs, the initial reservoir pressure is higher than the critical pressure of CO_2 (7370 kPa), so that the phase of injected $CO₂$ is in a supercritical condition except that in the Halfmoon field [\[146\]](#page-11-17). Under the reservoir temperature, the highest saturation pressure in the pilot tests is 7102 kPa, which is much lower than the initial reservoir pressure. Thus, when CO₂ is injected into the reservoir, the heavy oil in the transition zone between the injected $CO₂$ and the heavy oil is saturated. This results in gas drive and foamy oil flow occurring in the production stage (with the pressure decline process).

The reservoir depth of the successful cases applied in the oil fields ranges from 350 m to 1985 m with a thickness of net pay zone ranging from 12.2 m to 60 m. When the $CO₂$ huff 'n' puff process is applied in the heavy oil reservoir, the water saturation can be as high as 42.7%, which means that this process can be applied in a reservoir with a high water saturation (as proven in the lab tests) or even in a reservoir which has undergone the steam stimulation process [\[148\]](#page-11-18). Furthermore, with CO ² injection, the water cut of the produced fluids can be reduced signi ficantly [\[78,146,150\]](#page-10-5) .

The reservoir properties and operation parameters, in terms of reservoir connections, reservoir homogeneity, amount of $CO₂$ injection, soaking time, etc., affect the production performance of the CO₂ huff 'n' pu ff process applied in the heavy oil reservoir, signi ficantly. In regards to reservoir properties, the following aspects were mainly studied. The effect of the thickness of the net pay zone is sensitive to the production performance of the CO₂ huff 'n' puff process applied in a light oil reservoir [\[78,151,152\]](#page-10-5). However, in the heavy oil reservoir, the study indicated a di fferent trend for the e ffect of the thickness of the net pay zone [\[65\]](#page-10-37). The evaluation of the Forest Reserve oil field shows that the cumulative oil production does not increase with increases in the thickness of the net pay zone. In fact, it even decreases a little bit. The connections among the wells in the reservoir show a negative e ffect for the heavy oil production performance under the CO₂ huff 'n' puff process $[146]$. With well connection, the injected $CO₂$ will be pushed to the other wells nearby, which means that breakthrough will occur among di fferent wells, leading to the injected CO ² being pushed far away from the treated well. Thus, the pressure around the wellbore would not be maintained and the injected $CO₂$ would not be saturated in the heavy oil. Therefore, the oil production cannot be increased remarkably.

Another negative e ffect of the reservoir property is gas blockage near the wellbore, which forms another obstacle to the heavy oil pro-duction process [\[145,147\]](#page-11-19). In some wells, when $CO₂$ is injected into the reservoir, the injected CO_2 could not be pushed away from the vicinity of the wellbore due to unfavorable permeability around the wellbore. The injected CO ² could not be dissolved into heavy oil in a further location. Consequently, in the production stage, the pressure in the reservoir declined rapidly and the production of heavy oil could not be improved too much.

For the operation parameters, the injection volume and injection rate were mainly focused on. The cumulative oil production increased

Table 4

Properties of the heavy oil reservoirs those performed $CO₂$ huff 'n' puff pilot tests.

Table 5

Production performance of heavy oil by using $CO₂$ huff 'n' puff process in the pilot tests.

The Number of wells is based on the details studied in the literature.

 1 The average injection and production data are based on all the tested wells.

² The average injection and production data are related to each tested well in the studied area.

³ The average production rates are various: natural flow rate at 100 Stb/D; flow rate with pump at 50–60 Stb/D; then stabilized at 25–30 Stb/D.

⁴ The injection and production data are based on all the three cycles of the studied well.

with greater $CO₂$ volume injected into the reservoir [\[65\]](#page-10-37). Because a larger volume of $CO₂$ was injected into the reservoir, a larger area of connections between $CO₂$ and heavy oil occurred, leading to more heavy oil being saturated with $CO₂$ and the production of more heavy oil during the production stage. Even under the same injection volume, the injection rate of $CO₂$ can affect the heavy oil production performance to some extent. With a higher injection rate, the heavy oil recovery is greater, because the higher $CO₂$ injection rate promotes more serious $CO₂$ fingering in the heavy oil reservoir than a lower injection rate $[153, 154]$. Consequently, the injected $CO₂$ would be pushed into a further area in the reservoir and enlarge the connection area.

[Table 5](#page-7-1) shows that the stimulation ratio ranges from 4.2 to 18 MScf/ Stb. The stimulation ratio is defined as the peak oil production rate after the $CO₂$ injection process divided by that before $CO₂$ injection [\[65,155\]](#page-10-37). Thus, the $CO₂$ huff 'n' puff process applied in the heavy oil reservoir is effective and economical. The reservoir properties and operation parameters for a successful $CO₂$ huff 'n' puff process applied in the heavy oil reservoir can be summarized as: a thick oil pay zone, deep reservoir, mild pressure support, an approximate 2–4 weeks soaking time, a high $CO₂$ injection rate and volume, and a maximum of 3 cycles [\[65\]](#page-10-37).

6. Numerical and mathematical studies

The numerical and mathematical studies on the $CO₂$ huff 'n' puff process in heavy oil reservoirs in the published literature focus mainly on numerical simulation by using commercial simulators and mathematical study through the correlation method. Regarding the numerical simulation, different simulators were applied to conduct history match studies in terms of a single well simulator [\[156\],](#page-11-24) dual-porosity simulator [\[157\],](#page-11-25) thermal 3-D and 3-phase simulator [\[148\],](#page-11-18) compositional simulator [\[40\]](#page-9-20), and E300 [\[144\]](#page-11-22). The simulation results indicate that:

- (1) The $CO₂$ huff 'n' puff process could be conducted to enhance oil production in heavy oil reservoir, successfully;
- (2) The effect of the reservoir properties shows that the drainage area would affect the $CO₂$ huff 'n' puff process due to this process only impacting a certain area near the wellbore. If the drainage area is greater than the certain area, the incremental oil would not increase much;
- (3) The recommended application of the $CO₂$ huff 'n' puff process differs according to the viscosity of the heavy oil. For the extra heavy oil (with an oil viscosity greater than 10,000 mPa·s), one or two cycles of the CSS process in the early stage will benefit $CO₂$ huff 'n' puff stimulation. For the regular heavy oil reservoir (with a viscosity ranging from 2000 to 10,000 mPa·s), even three to four cycles of the CSS process will benefit it. In addition, the $CO₂$ stimulation process can obtain oil production. For the heavy oil reservoir (with a viscosity lower than 500 mPa·s), the $CO₂$ huff 'n' puff process should be carried out directly [\[148\]](#page-11-18);
- (4) The effect of operation parameters indicate that (a) an optimized soaking time (24 h) exists in the liquid phase $CO₂$ huff 'n' puff process; (b) a lower oil recovery factor is obtained when a higher injection rate is applied (the higher injection rate will push the oil near the producer to a further area of the core. Moreover, the higher injection rate leads to a shorter duration for the connection of heavy oil and liquid $CO₂$); and (c) a higher oil recovery factor is gained when a larger volume of liquid phase $CO₂$ is injected due to the larger volume of heavy oil being connected with the injected $CO₂$.

For the correlation method, the effect of the key parameters including operation parameters (treatment pressure, treatment volume, backpressure, and the number of cycles) and reservoir properties (oil viscosity, reservoir depth, and current oil saturation) on the $CO₂$ huff 'n' puff process in heavy oil reservoirs were studied [\[158\],](#page-11-26) and an analytical method was proposed to predict heavy oil production in the oil field [\[159\]](#page-11-29). The scholars studied the relationship among the reservoir properties, operation parameters, and production efficiency. Regarding the operation parameters, (1) a higher treatment pressure (the maximum reservoir pressure permitted during the injection process) can gain lower oil viscosity, because more $CO₂$ can be injected into the reservoir. The injection pressure can be as high as 15.8 kPa/m of depth, which was carried out in several fields with good heavy oil production performance; (2) backpressure may benefit the oil production, and the productivity increases with a declining bottom hole pressure. Regarding the reservoir properties, (1) commercial projects indicate that the heavy oil viscosity is usually less than 2000 mPa·s; (2) with a higher depth, the reservoir can obtain a higher injection pressure, which results in a higher $CO₂$ solubility and lower oil viscosity; (3) unexpectedly, high oil saturation tends to reduce the performance of the $CO₂$ huff 'n' puff process due to the incremental production being reduced. A correlation was developed for the efficiency of heavy oil production and the parameters [\[158\]:](#page-11-26)

$$
E = 0.33 - 0.035N_c - 4.5 \times 10^{-5} \mu_o + 1.6 \times 10^{-4} P_t + 1.3 \times 10^{-9} P_t^2
$$

+ 4.3 × 10⁻⁵k - 0.013S_{oi} - 0.69V_c

where, *E* is the efficiency of the CO₂ huff 'n' puff process, STB oil/MScf CO₂; N_c is the number of cycles; μ_o is the heavy oil viscosity, mPa·s; P_t is the treatment pressure, kPa; k is the reservoir permeability, mD; S_{oi} is the initial oil saturation, fraction; and V_c is the volume of injected CO_2 in each cycle per foot of sand, MMSc/ft.

The heavy oil production rate, which relates to the pressure distribution, heavy oil, and reservoir properties, was correlated according to the production data in the oil fields. To simplify the calculation, the velocities of CO₂ diffusion and heat transfer are assumed as constants. The pressure distribution was calculated using Spivey's model. An exact solution was obtained and the temperature distribution was calculated using Laplace's transformation based on Marx and Langenheim's model [\[160\].](#page-11-30) Carbon dioxide concentration in the transient zone is an important parameter to predict the location of $CO₂$ dissolved into heavy oil and how much heavy oil can be produced in each cycle, and Fick's law diffusion was applied to calculate $CO₂$ concentration in the transient zone. Then the oil production rate was predicted based on Boberg and Lantz's model $[161]$, and the heavy oil production rate in the $CO₂$ huff 'n' puff process was presented as [\[159\]](#page-11-29):

$$
q_o = \frac{0.536 \cdot h_t \cdot K_o \cdot (P_{(r,t)} - P_{wf})}{B_o \cdot \mu_{oc} \cdot \ln\left(\frac{r_L}{r_{md}}\right) + B_{o - CO_2} \cdot \mu_{oh} \cdot \ln\left(\frac{r_{md}}{r_{vo}}\right)}
$$

where, q_o is the heavy oil production rate, bbl/day; h_t is the reservoir thickness, m; K_o is the oil relative permeability, mD; $P_{(r,t)}$ is the pressure as a function of space and time, kPa; P_{wf} is the wellbore flowing pressure, kPa; B_0 is the heavy oil volume factor, rb/stb; μ_{oc} is the viscosity of heavy oil-CO₂ system, mPa·s; r_L is the limit radius, m; r_{md} is the maximum diffusion radius, m; B_{0-CO_2} is the volume factor of heavy oil-CO₂ system, rb/stb; r_w is the well radius, m.

7. Challenges of CO₂ huff 'n' puff process

Through worldwide application in heavy oil reservoirs, $CO₂$ has been found to be an efficient candidate for recovering heavy oil in low pressure and thin pay zone reservoirs via the immiscible displacement process [\[162,163\]](#page-11-32). Although the $CO₂$ huff 'n' puff process has been carried out in the field for several decades, technic and economic challenges are still encountered when it is implemented in the field. Technically, asphaltene precipitation, corrosion, viscous fingering, etc., are the main serious problems. Economically, oil prices and the cost of $CO₂$ recourse are the key obstacles.

Asphaltene deposition in the $CO₂$ huff 'n' puff process causes serious issues such as formation damage, relative permeability reduction, and flow interruption in the reservoir and the surface facilities. It can lead to

a low production rate and even no flow rate when the tubes or the wellbore are plugged [\[75,164](#page-10-39)–168]. Furthermore, when the asphaltene content is higher than 4.6%, the wettability of the reservoir will alternate from water wet to oil wet, which results in a reduction in heavy oil production $[169,170]$. To reduce asphaltene deposition in the CO₂ huff 'n' puff process, asphaltene inhibitors can be injected into the reservoir. The results of the tests show that the inhibitor can effectively reduce the asphaltene deposition in the gas lift production process in the heavy oil production [\[171\]](#page-11-34). Another way to reduce the asphaltene deposition in heavy oil is by adding a solvent mixture of polar protic hexane-1-ol and nonpolar toluene into the heavy oil along with the $CO₂$ injection process [\[172\]](#page-11-35). The tests indicate that the asphaltene deposition can be delayed or stopped when the solvents are injected.

In the production stage, the undissolved $CO₂$ evolves out of the reservoir through the penetrated holes on the well. The volume of the $CO₂$ expands extremely in a very short time, which results in heat extraction at the bottom of the wellbore. When the temperature decreases to the wax appearance temperature, wax precipitation occurs in the heavy oil at the bottom of the well, the wax sticks on the wellbore and the formation near the wellbore [\[148\]:](#page-11-18) leading to the flow rate decreasing [\[173\]](#page-11-36). The efficient method to reduce wax precipitation in the reservoir is to optimize the pressure depletion rate, under which the temperature near the wellbore would not be decreased too much and the heavy oil production rate would not be affected too much. Another approach is to add a wax inhibitor into the reservoir. The polymeric inhibitor causes the formation of a hydrocarbon chain between the wax inhibitor and the wax. The chain is a polar segment which inhibits the aggregation stage of the wax, so the wax appearance time can be decreased [\[174\].](#page-11-37)

The corrosion of equipment due to corrosive fluid is generated when the injected $CO₂$ encounters water. The chloride corrosion is serious in the $CO₂$ huff 'n' puff process implemented in the Bati Raman oilfield and Aminoil's North Bolsa Strip project, even though special steel and chemical protection is used $[156,175]$. The best way to avoid CO₂ acid corrosion is to add inhibitor chemicals into the reservoir and use a corrosion resistant element on the surface of the pipeline, metal components, etc. [\[176\].](#page-11-38)

Because the viscosity of $CO₂$ is much lower than that of the reservoir fluids (heavy oil, formation water), viscous fingering occurs when $CO₂$ is injected into the reservoir. The injected $CO₂$ will pass through the higher permeability zone and bypass the lower permeability zone. In the oil field, injected $CO₂$ breakthrough occurs among nearby wells in a short time in the heterogeneous reservoirs [177–[179\].](#page-11-39) For this type of reservoir, high permeability layer(s) exist and the injected $CO₂$ passes through these layers, except they will be maintained near the injector.

Different methods can be used to avoid the negative effect of high permeability layers among wells in different oil pay zones. First, a packer can be applied to isolate thin high permeability layers in the wellbore [\[146\]](#page-11-17). For a thick layer or a reservoir with natural fractures, which means there is a large amount of heavy oil reserves in this layer, the packer is not applicable. To solve this problem, a high viscosity gel solution or another type of solution can be injected into the reservoir to separate the high permeability layer prior to the implementation of the CO2 huff 'n' puff process. By using this method, heavy oil production can be improved [\[145\]](#page-11-19).

Other technical issues such as low injectivity, pump problems, ice plugs forming in the injection lines and tubes, and annular wellbore freezing occurring while the injection pressure is extremely low, etc., are also serious problems in the field [\[49,53\]](#page-10-40).

In terms of the economic aspect, the costs of $CO₂$ capture, $CO₂$ transportation, facilities of $CO₂$ injection system, operation etc., are the main capital investments. Until now, successful field applications of the $CO₂$ huff 'n' puff process have been based on $CO₂$ reservoirs located close to the heavy oil reservoirs. For example, the $CO₂$ source for the CO2 huff 'n' puff process applied in the Bati Raman field, Turkey, is a $CO₂$ reservoir (the Dodan field) located 88 km away from the heavy oil reservoir. However, regarding the application of the process without a $CO₂$ reservoir nearby, the benefits of the $CO₂$ huff 'n' puff process are not remarkable, because the costs of $CO₂$ capture and transportation without pipelines remain very high. Furthermore, the facilities used in the $CO₂$ huff 'n' puff process are specialized from common ones to avoid the corrosion of $CO₂$. In addition, the operations on the $CO₂$ huff 'n' puff field are much more complex than other common processes.

8. Conclusion

Five conclusions can be drawn from this study. First, the experimental, pilot tests and numerical studies show that the $CO₂$ huff 'n' puff process is efficient to enhance oil recovery in heavy oil reservoirs.

Second, the main challenges of the $CO₂$ huff 'n' puff process include asphaltene precipitation, wax deposition, equipment corrosion, viscous fingering, etc. were analyzed, and the solutions for the potential problems have been researched in detail.

Third, viscosity reduction is extremely remarkable in the heavy oil- $CO₂$ system, and the viscosity reduction ratio can reach up to 99.8%. Oil swelling is significant in the $CO₂$ -based EOR process, and the oil swelling factor can be as high as 1.28 when $CO₂$ is recombined into heavy oil.

Fourth, the $CO₂$ huff 'n' puff process is one of the most successful EOR methods in heavy oil production from both experimental studies and field tests. The property range of the heavy oil in which the $CO₂$ huff 'n' puff process can be used to enhance heavy oil production is large. The oil viscosity can be as high as 28,646 mPa·s, the oil gravity can be as low as 4 °API, and the $CO₂$ utilization can be as low as 4.2 MScf/Stb.

Fifth, in terms of the reservoir properties of the heavy oil reservoir that can implement the $CO₂$ huff 'n' puff process successfully, this study indicates that the depth can be as high as 1985 m, the oil pay zone can be as low as 12.2 m, the permeability can range from 30 to 24,200 mD, and the porosity can range from 11% to 37.5%.

Acknowledgments

The authors would like to acknowledge the National Natural Science Foundation of China-China (Key Program) (Grant No. 51534006) and the Natural Science Foundation of China-China (Grant No. 51374181) for their technical and financial supports.

References

- [1] [Briggs PJ, Baron RP, Fulleylove RJ, Wright MS. Development of heavy-oil re](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0005)[servoirs. J Pet Technol 1988;40\(2\):206](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0005)–14.
- [2] [Meyer RF, Attanasi ED. Heavy oil and natural bitumen](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0010)—strategic petroleum re[sources. U.S. Geological Survey; 2003.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0010)
- [3] PetroWiKi, "Heavy Oil"; 2016.
- [4] [Jiang T. An improved vapour solvent injection technique for enhanced heavy oil](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0020) [recovery. University of Regina; 2013.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0020)
- [5] Safi[nya K. Heavy oil recovery: the road ahead. Alberta Oil 2008;4\(1\):15](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0025)–20.
- [6] [Santos RG, Loh W, Bannwart AC, Trevisan OV. An overview of heavy oil properties](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0030) [and its recovery and transportation methods. Braz J Chem Eng 2014;31\(3\):571](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0030)–90.
- [7] [Zhou X. Experimental study on foamy oil](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0035) flow by using different heavy oil-solvent [systems. University of Regina; 2015.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0035)
- [8] Wang H, Zeng F, Zhou X. Study of the Non-Equilibrium PVT Properties of Methaneand Propane-Heavy Oil Systems. In: SPE Canada Heavy Oil Technical Conference; 2015.
- [9] [Zhou X, Zeng F, Zhang L, Wang H. Foamy oil](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0045) flow in heavy oil-solvent systems [tested by pressure depletion in a sandpack. Fuel 2016;171:210](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0045)–23.
- [10] [Butler R. SAGD comes of AGE. J Can Pet Technol 1998;37\(7\):9](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0050)–12. [Morte M, Hascakir B. Estimation of pseudo-relative permeability curves through](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0055) [an analytical approach for steam assisted gravity drainage \(SAGD\) and solvent](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0055)
- [aided-steam assisted gravity drainage. J Unconv Oil Gas Resour 2016;16:45](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0055)–52. [12] [Zhou X, Zeng F, Zhang L. Improving Steam-Assisted Gravity Drainage performance](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0060) [in oil sands with a top water zone using polymer injection and the](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0060) fishbone well
- [pattern. Fuel 2016;184:449](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0060)–65. [13] [Yoelin SD. The TM sand steam stimulation project. J Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0065)
- [1971;23\(8\):987](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0065)–94.
- [14] [Bao Y, Wang J, Gates ID. On the physics of cyclic steam stimulation. Energy](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0070) [2016;115:969](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0070)–85.
- [15] [Owens HE, Bramlsy BG. Performance of equipment used in high-pressure steam](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0075) fl[oods. J Pet Technol 1966;18\(12\):1525](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0075)–31.
- [16] [Liu P, Zheng H, Wu G. Experimental study and application of steam](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0080) flooding for [horizontal well in ultraheavy oil reservoirs. J Energy Resour Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0080) [2016;139\(1\):1](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0080)–9. p. 012908.
- [17] Jones [RF, Truitt NE. In situ combustion a way from thin, horizontal gas channels.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0085) [Soc Pet Eng J 1968;8\(1\):18](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0085)–32.
- [18] [Moore RG, Laureshen J, Ursenbach MG, Mehta SA. In situ combustion reservoirs in](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0090) [Canadian heavy oil. Fuel 1995;74\(8\):1169](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0090)–75.
- [19] [Jia N, Law DH, Naccache P, Giddins MA. An investigation of the applicability of](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0095) [kinetic models for in-situ combustion processes with di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0095)fferent types of oils. Nat [Resour Res 2017;26\(1\):37](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0095)–59.
- [20] [Ivory J, Chang J, Coates R, Forshner K. Investigation of cyclic solvent injection](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0100) [process for heavy oil recovery. J Can Pet Technol 2010;49\(9\):22](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0100)–33.
- [21] [Jiang T, Zeng F, Jia X, Gu Y. A new solvent-based enhanced heavy oil recovery](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0105) [method: cyclic production with continuous solvent injection. Fuel](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0105) [2014;115:426](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0105)–33.
- [22] [Das SK, Butler RM. Mechanism of the vapor extraction process for heavy oil and](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0110) [bitumen. J Pet Sci Eng 1998;21\(1](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0110)–2):43–59.
- [23] [Ma H, Yu G, She Y, Gu Y. A parabolic solvent chamber model for simulating the](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0115) [solvent vapor extraction \(VAPEX\) heavy oil recovery process. J Pet Sci Eng](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0115) [2017;149:465](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0115)–75.
- [24] Sayegh SG, Maini BB. Laboratory evaluation of the $CO₂$ huff-N-puff process for [heavy oil reservoirs. J Can Pet Technol 1984;23\(3\):29](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0120)–36.
- [25] Lu T, Li Z, Fan W, Li S. CO₂ huff and puff [for heavy oil recovery after primary](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0125) [production. Greenh Gases Sci Technol 2016;6:288](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0125)–301.
- [26] [Srivastava R, Huang S, Dong M. Comparative e](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0130)ffectiveness of $CO₂$ produced gas, and fl[ue gas for enhanced heavy-oil recovery. SPE Reserv Eval Eng](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0130) [1999;2\(3\):238](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0130)–47.
- [27] [Bowers B, Drummond KJ. Conventional crude oil resources of the western Canada](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0135) [sedimentary basin. J Can Pet Technol 1997;36\(2\):56](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0135)–63.
- [28] Huang SS, Pappas ES, Jha KN. The carbon dioxide immiscible recovery process and its potential for saskatchewan reservoirs. In: Technical Meeting/Petroleum Conference of the South Saskatchewan Section, October 6–8, Regina; 1987.
- [29] [Zhao DW, Wang J, Gates ID. Thermal recovery strategies for thin heavy oil re](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0145)[servoirs. Fuel 2014;117:431](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0145)–41.
- [30] [Srivastava RK, Huang SS, Mourits FM. A laboratory evaluation of suitable oper](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0150)[ating strategies for enhanced heavy oil recovery by gas injection. J Can Pet](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0150) [Technol 1997;36\(2\):33](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0150)–41.
- [31] Shi R, Kantzas A. An investigation of oil viscosity and depletion rate effect on heavy oil recovery by primary production and CO₂ huff and puff process. In: CSPG CSEG CWLS Convention; 2008, p. 480–483.
- [32] [Tharanivasan AK, Yang C, Gu Y. Measurements of molecular di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0160)ffusion coefficients [of carbon dioxide, methane, and propane in heavy oil under reservoir conditions.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0160) [Energy Fuels 2006;20:2509](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0160)–17.
- [33] Yang C, Gu Y, Diffusion coeffi[cients and oil swelling factors of carbon dioxide,](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0165) [methane, ethane, propane, and their mixtures in heavy oil. Fluid Phase Equilib](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0165) [2006;243:64](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0165)–73.
- [34] Upreti SR, Mehrotra AK. Diffusivity of CO_2 , CH_4 , C_2H_6 and N_2 in Athabasca bi[tumen. Can J Chem Eng 2002;80\(February\):116](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0170)–25.
- [35] [Du Z, Zeng F, Chan C. An experimental study of the post-CHOPS cyclic solvent](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0175) [injection process. J Energy Resour Technol 2015;137\(4\):42901.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0175)
- [36] [Butler R, Jiang Q. Improved recovery of heavy oil by Vapex with widely spaced](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0180) [horizontal injectors and producers. J Can Pet Technol 2000;39\(1\):48](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0180)–56.
- [37] Karmaker K, Maini B. Experimental investigation of oil drainage rates in the vapex process for heavy oil and bitumen reservoirs. In: SPE Annual Technical Conference and Exhibition; 2003.
- [38] [Butler RM, Mokrys L. Recovery of heavy oils using vapourized hydrocarbon sol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0190)[vents: further development of the Vapex process. J Can Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0190) [1993;32\(6\):56](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0190)–62.
- [39] [Cuthiell D, McCarthy C, Frauenfeld T, Cameron S, Kissel G. Investigation of the](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0195) [VAPEX process using ct scanning and numerical simulation. J Can Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0195) [2003;42\(2\):41](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0195)–9.
- [40] Ravel F, Anterion F. Numerical simulation of $CO₂$ -heavy oil interactions in fractured medium: an interfacial film concept. In: SPE Annual Technical Conference and Exhibition, 22–26 September, Las Vegas, Nevada; 1985.
- [41] Alshmakhy A, Maini B. A follow-up recovery method after cold heavy oil production cyclic CO₂ injection. In: Proceedings of SPE Heavy Oil Conference Canada; 2012.
- [42] Solomon S, et al. Climate change 2007: The Physical Science Basis. Contribution of Working Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change Group I to the Fourth Assessment Report of the Intergovernmental Panel on Climate Change. Cambridge, UK: Cambridge University Press; 2007.
- [43] [Schrag DP. Preparing to capture carbon. Science \(80-\) 2007;315\(5813\):812](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0215)–3.
- [44] [Perera M, Ranjith P, Choi SK, Bouazza A, Kodikara J, Airey D. A review of coal](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0220) [properties pertinent to carbon dioxide sequestration in coal seams: with special](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0220) [reference to Victorian brown coals. Environ Earth Sci 2011;64:223](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0220)–35.
- [45] Huntingford C, et al. Simulated resilience of tropical rainforests to CO_2 -induced [climate change. Nat Geosci 2013;6:268](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0225)–73.
- [46] [Seinfeld JH, Pandis SN. Atmospheric chemistry and physics: from air pollution to](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0230) [climate change. Hoboken, New Jersey: John Wiley & Sons Inc; 2016.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0230)
- [Silva LCR, Anand M. Probing for the in](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0235)fluence of atmospheric $CO₂$ and climate [change on forest ecosystems across biomes. Glob Ecol Biogeogr 2013;22\(1\):83](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0235)–92.
- [48] Chang J, et al. Effect of climate change, $CO₂$ [trends, nitrogen addition, and land](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0240)[cover and management intensity changes on the carbon balance of European](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0240)

[grasslands. Glob Change Biol 2016;22\(1\):338](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0240)–50.

- [49] Gao C, Li X, Guo L, Zhao F. Heavy oil production by CO₂ injection. Greenh Gases [Sci Technol 2012;2\(6\):408](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0245)–18.
- [50] Rojas G, Farouq Ali S. Dynamics of subcritical CO₂/brine floods for heavy-oil re[covery. SPE Reserv Eng 1988;3\(1\):35](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0250)–44.
- Jha K. A laboratory study of heavy oil recovery with $CO₂$ injection. J Can Pet [Technol 1986.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0255)
- [52] Holm L, O'[Brien L. Carbon dioxide test at the mead-strawn](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0260) field. J Pet Technol [1971;23\(4\):431](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0260)–42.
- [53] [Sahin S, Kalfa U, Celebioglu D. Bati Raman](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0265) field immiscible $CO₂$ application[status quo and future plans. SPE Reserv Eval Eng 2008;11\(4\):778](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0265)–91.
- [54] Rogers DW. Concise physical chemistry. Brooklyn; 2011.
- [55] Roche P. Cheap, green, heavy oil; 2017.
[56] Simpson MR. The CO₂ buff 'n' puff proce
- Simpson MR. The $CO₂$ huff 'n' puff [process in a bottomwater-drive reservoir. J Pet](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0280) [Technol 1988;40\(7\):887](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0280)–93.
- [57] Murray MD, Frailey SM, Lawal AS. New approach to $CO₂$ flood: soak alternating gas. In: SPE Permian Basin Oil and Gas Recovery Conference; 2001.
- [58] Mangalsingh D, Jagai T. A laboratory investigation of the carbon dioxide. In: SPE Latin America/Caribbean Petroleum Engineering Conference; 1996.
- [59] Spivak A, Chima CM. Mechanisms of immiscible C02 injection in heavy oil reservoirs, Wilmington Field, CA. In: SPE Enhanced Oil Recovery Symposium, 15–18 April, Tulsa, Oklahoma; 1984.
- [60] Abedini A, Torabi F. On the $CO₂$ [storage potential of cyclic CO2 injection process](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0300) [for enhanced oil recovery. Fuel 2014;124:14](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0300)–27.
- [61] Zhang Y, Sayegh SG, Huang S. Improved heavy oil recovery by CO₂ injection augmented with chemicals. In: International Oil and Gas Conference and Exhibition in China; 2010.
- [62] Beeson DM, Ortloff [GD. Laboratory investigation of the water-driven carbon di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0310)[oxide process for oil recovery. J Pet Technol 1959;11\(4\):63](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0310)–6.
- [63] Holm LW. Carbon dioxide solvent fl[ooding for increased oil recovery. Pet Trans](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0315) [AIME 1959;216:225](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0315)–31.
- [64] Khatib AK, Earlougher RC, Kantar K, $CO₂$ injection as an immiscible application for enhanced recovery in heavy oil reservoirs. In: SPE California Regional Meeting, 25-27 March, Bakersfield, California; 1981.
- [65] Mohammed-Singh LJ, Singhal AK, Sim SS-K. Screening Criteria for CO₂ Huff 'n' Puff Operations. In: SPE/DOE Symposium on Improved Oil Recovery; 2006.
- [66] Gondiken S. Camurlu field immiscible CO₂ huff and puff pilot project. In: Middle East Oil Show; 1987.
- [67] Jeong MS, Lee KS. Maximizing oil recovery for $CO₂$ huff and puff process in pilot scale reservoir. In: Advances in Civil, Environmental, and Materials Research (ACEM15); 2015.
- [68] Firouz AQ, Torabi F. Feasibility study of solvent-based huff-n-puff method (cyclic solvent injection) to enhance heavy oil recovery. In: SPE Heavy Oil Conference Canada, 12-14 June, Calgary, Alberta, Canada; 2012.
- [69] Miller J, Jones R. A laboratory study to determine physical characteristics of heavy oil after CO₂ saturation. In: SPE/DOE Enhanced Oil Recovery Symposium, 5-8 April, Tulsa, Oklahoma, 1981.
- [70] Klins MA. Carbon dioxide fl[ooding: basic mechanisms and project design. 1st ed.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0350) [Boston: D. Reidel Publishing Company; 1984.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0350)
- [71] [Bijeljic B, Muggeridge AH, Blunt MJ. Multicomponent mass transfer across water](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0355) fi[lms during hydrocarbon gas injection. Chem Eng Sci 2003;58\(11\):2377](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0355)–88.
- [72] [Maneeintr K, Babadagli T, Sasaki K, Sugai Y. Analysis of heavy oil emulsion-carbon](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0360) [dioxide system on oil-swelling factor and interfacial tension by using pendant drop](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0360) [method for enhanced oil recovery and carbon dioxide storage. Int J Environ Sci](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0360) [Dev 2014;5\(2\):118](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0360)–23.
- [73] Seyyedsar SM, Farzaneh SA, Sohrabi M. Investigation of low-density CO_2 injection [for enhanced oil recovery. Ind Eng Chem Res 2017;56\(18\):5443](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0365)–54.
- [74] Hwango J, Park SJ, Deo MD, Hanson FV. Phase behavior of $CO₂/crude$ oil mixtures in supercritical fl[uid extraction system: experimental data and modeling. Ind Eng](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0370) [Chem Res 1995;34\(4\):1280](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0370)–6.
- [75] Vazquez D, Mansoori GA. Identifi[cation and measurement of petroleum pre](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0375)[cipitates. J Pet Sci Eng 2000;26:49](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0375)–55.
- [76] Huang T, Zhou X, Yang H, Liao G, Zeng F. CO₂ flooding strategy to enhance heavy [oil recovery. Petroleum 2017;3:68](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0380)–78.
- [77] [Wu K, Chen Z, Li J, Li X, Xu J, Dong X. Wettability e](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0385)ffect on nanoconfined water fl[ow. Proc Natl Acad Sci USA 2017;114\(13\):3358](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0385)–63.
- [78] Monger TG, Ramos JC, Thomas J. Light oil recovery from cyclic $CO₂$ injection: influence of low pressures, impure CO₂, and reservoir gas. SPE Reserv Eng [1991;6\(1\):25](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0390)–32.
- [79] [Sun X, Zhang Y, Chen G, Gai Z. Application of nanoparticles in enhanced oil re](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0395)[covery: a critical review of recent progress. Energies 2017;10\(3\):345.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0395)
- [80] Li X, Yin Y, Yang Z, Shen Y, Liu Z. Produced gas reinjection based cyclic solvent processes for foamy oil reservoirs in the eastern Orinoco belt, Venezuela. In: SPE Canada Heavy Oil Technical Conference, 7-9 June, Calgary, Alberta, Canada; 2016.
- [81] [Or C, Sasaki K, Sugai Y, Nakano M, Imai M. Swelling and viscosity reduction of](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0405) heavy oil by CO₂-gas foaming in immiscible condition. SPE Reserv Eval Eng [2016;19\(2\).](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0405)
- [82] [Firouz AQ, Torabi F. Utilization of carbon dioxide and methane in hu](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0410)ff-and-puff [injection scheme to improve heavy oil recovery. Fuel 2014;117:966](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0410)–73.
- [83] [Sheng JJ, Hayes RE, Maini B, Tortike W. Modelling foamy oil](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0415) flow in porous [media. Transp Porous Media 1999;35:227](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0415)–58.
- Maini BB, Sarma HK, George AE. Signifi[cance of foamy-oil behaviour in primary](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0420) [production of heavy oils. J Can Pet Technol 1993;32\(9\):50](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0420)–4.
- [85] [Zhou X, Yuan Q, Zeng F, Zhang L, Jiang S. Experimental study on foamy oil be](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0425)[haviour using a heavy oil-methane system in the bulk phase. J Pet Sci Eng](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0425)

[2017;158:309](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0425)–21.

- [86] Asghari K, Torabi F. Laboratory experimental results of huff-and-puff $CO₂$ flooding in a fractured/nCore system. In: Proceedings of SPE Annual Technical Conference and Exhibition; 2007.
- [87] Zhou X, Zeng F, Wang H, Hong SY. Study on foamy oil production performance by using different solvents in laboratory. In: 35th workshop & Symposium IEA Collaborative Project on EOR; 2014.
- [88] [Du Z, Zeng F, Peng X, Chan C. Optimizing the pressure decline rate on the cyclic](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0440) [solvent injection process for enhanced heavy oil recovery. J Pet Sci Eng](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0440) [2016;145:629](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0440)–39.
- [89] [Ostos AN, Maini BB. An integrated experimental study of foamy oil](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0445) flow during [solution gas drive. J Can Pet Technol 2005;44\(4\):43](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0445)–50.
- [90] Zhang Y. Eff[ects of temperature on foamy solution gas-drive. University of Calgary;](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0450) [1999.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0450)
- [91] [Tang G-Q, Firoozabadi A. Gas- and liquid-phase relative permeabilities for cold](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0455) [production from heavy-oil reservoirs. SPE Reserv Eval Eng 2003;6\(2\).](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0455)
- [92] [Liu P, Wu Y, Li X. Experimental study on the stability of the foamy oil in devel](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0460)[oping heavy oil reservoirs. Fuel 2013;111:12](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0460)–9.
- [93] [Dyer S, Huang S, Farouq Ali S, Jha K. Phase behaviour and scaled model studies of](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0465) [prototype saskatchewan heavy oils with carbon dioxide. J Can Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0465) [1994;33\(8\):42](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0465)–8.
- [94] Jeff[ries-Harris MJ, Coppel C. Solvent stimulation in low gravity oil reservoirs. J Pet](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0470) [Technol 1969;21\(2\):167](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0470)–75.
- [95] Khatib AK, Earlougher RC. $CO₂$ injection as an immiscible application for enhanced recovery in heavy oil reservoirs. In: SPE California Regional Meeting; 1981.
- [96] [Chung FTH, Jones RA, Nguyen HT. Measurements and correlations of the physical](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0480) properties of CO₂/heavy-crude-oil mixtures. PE Reserv Eng 1988;3(3):822-8.
- [97] Bagci AS, Olushola S, Mackay E. Performance analysis of SAGD wind-down process with CO₂ injection. In: SPE/DOE Improved Oil Recovery Symposium; 2008.
- [98] [Frauenfeld T, Lillico D, Jossy C, Vilcsak G, Rabeeh S, Singh S. Evaluation of par](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0490)[tially miscible processes for Alberta heavy oil reservoirs. J Can Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0490) [1998;37\(4\):17](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0490)–24.
- [99] Seyyedsar SM, Farzaneh SA, Sohrabi M. Enhanced heavy oil recovery by intermittent CO₂ injection. In: SPE Annual Technical Conference and Exhibition, 28-30 September, Houston, Texas, USA; 2015.
- [100] [Sayegh SG, Rao DN, Kokal S, Najman J. Phase behaviour and physical properties](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0500) of Lindbergh heavy oil/CO₂ [mixtures. J Can Pet Technol 1990;29\(6\):31](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0500)-9.
- [101] [Jha KN. A laboratory study of heavy oil recovery with carbon dioxide. J Can Pet](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0505) [Technol 1986;25\(2\):54](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0505)–63.
- [102] Li X, et al. Produced gas reinjection based cyclic solvent processes for foamy oil reservoirs in the eastern Orinoco belt, Venezuela; 2016.
- [103] Zhang Y, Huang S, Luo P. Coupling immiscible CO₂ technology and polymer in[jection to maximize EOR performance for heavy oils. J Can Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0515) [2010;49\(5\):27](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0515)–33.
- [104] Srivastava RK, Huang SS, Dyer S, Mourits F. Heavy oil recovery by subcritical carbon dioxide flooding. In: SPE Latin America/Caribbean Petroleum Engineering Conference; 1994.
- [105] Zhang K. Research on heat transfer and viscosity reduction mechanism of $CO₂$ [assisted steam stimulation. China University of Petroleum; 2008.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0525)
- [106] [Zhou P. Study on reducing viscosity of super heavy crude oil with supercritical](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0530) CO₂. China University of Petroleum: 2010.
- [107] [Sankur V, Creek J, DiJulio S, Emanuel A. A laboratory study of Wilmington tar](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0535) zone $CO₂$ [injection project. SPE Reserv Eng 1986;1\(1\):95](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0535)-104.
- [108] Do H, Pinczewski W. Diff[usion controlled swelling of reservoir oil by direct contact](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0540) [with injection gas. Chem Eng Sci 1991;46\(5](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0540)–6):1259–70.
- [109] [Avaullee L, Neau E, Jaubert JN. Thermodynamic modeling for petroleum](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0545) fluid III. Reservoir fl[uid saturation pressures. A complete PVT property estimation.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0545) [Application to swelling test. Fluid Phase Equilib 1997;141\(1](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0545)–2):87–104.
- [110] [Welker JR, Dunlop DD. Physical properties of carbonated oils. J Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0550) [1963;15\(8\):873](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0550)–6.
- [111] [Grogan AT, Pinczewski WV. The role of molecular di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0555)ffusion processes in tertiary CO2 fl[ooding. J Pet Technol 1987;39\(5\):591](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0555)–602.
- [112] [Riazi MR. A new method for experimental measurement of di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0560)ffusion coefficients in reservoir fl[uids. J Pet Sci Eng 1996;14:235](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0560)–50.
- [113] [Wang L-S, Lang Z-X, Guo T-M. Measurement and correlation of the di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0565)ffusion coeffi[cients of carbon dioxide in liquid hydrocarbons under elevated pressures.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0565) [Fluid Phase Equilib 1996;117:364](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0565)–72.
- [114] Yuan Q, Zhou X, Zeng F, Knorr KD, Imran M. Effects of concentration-dependent diffusion on mass transfer and frontal instability in solvent-based processes. In: SPE Canada Heavy Oil Technical Conference; 2017.
- [115] Boustani A, Maini B. The role of diff[usion and convective dispersion in vapour](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0575) [extraction process. J Can Pet Technol 2001;40\(4\):68](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0575)–77.
- [116] [Frauenfeld T, Jossy C, Rispler K, Kissel G. Evaluation of the bottom water reservoir](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0580) [VAPEX process. J Can Pet Technol 2006;45\(9\):29](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0580)–35.
- [117] [Yuan Q, Zhou X, Zeng F, Knorr KD, Imran M. Nonlinear simulation of miscible](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0585) [displacements with concentration-dependent di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0585)ffusion coefficient in homogeneous [porous media. Chem Eng Sci 2017;172:528](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0585)–44.
- [118] Huang Y, Zhou X, Zeng F. Comparison study of two different methods on the localised Enkf on SAGD processes. In: International Petroleum Technology Conference; 2016.
- [119] [Ghasemi M, Astutik W, Alavian SA, Whitson CH. Determining di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0595)ffusion coeffi[cients for carbon dioxide injection in oil-saturated chalk by use of a constant](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0595)volume-diff[usion method. SPE J 2017;22\(2\):505](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0595)–20.
- [120] Jiang Q, Butler RM. Selection of well configurations in Vapex process. In: International Conference on Horizontal Well Technology; 1996.
- [121] [Tharanivasan AK, Yang C, Gu Y. Comparison of three di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0605)fferent interface mass [transfer models used in the experimental measurement of solvent di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0605)ffusivity in [heavy oil. J Pet Sci Eng 2004;44:269](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0605)-82.
- [122] [Kavousi A, Torabi F, Chan CW, Shirif E. Experimental measurement and para](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0610)metric study of $CO₂$ solubility and molecular diff[usivity in heavy crude oil systems.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0610) [Fluid Phase Equilib 2014;371:57](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0610)–66.
- [123] [Jamialahmadi M, Emadi M, Müller-Steinhagen H. Di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0615)ffusion coefficients of me[thane in liquid hydrocarbons at high pressure and temperature. J Pet Sci Eng](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0615) [2006;53\(1](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0615)–2):47–60.
- [124] [Zheng S, Li H, Sun H, Yang D. Determination of di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0620)ffusion coefficient for alkane solvent−CO2 [mixtures in heavy oil with consideration of swelling e](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0620)ffect. Ind Eng [Chem Res 2016;55\(6\):1533](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0620)–49.
- [125] [Li H, Yang D. Determination of individual di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0625)ffusion coefficients of solvent/CO₂ [mixture in heavy oil with pressure-decay method. SPE J 2016;21\(1\):131](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0625)–43.
- [126] [Upreti SR, Mehrotra AK. Experimental measurement of gas di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0630)ffusivity in bitumen: [results for carbon dioxide. Ind Eng Chem Res 2000;39:1080](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0630)–7.
- [127] [Sheikha H, Pooladi-Darvish M, Mehrotra AK. Development of graphical methods](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0635) for estimating the diffusivity coeffi[cient of gases in bitumen from pressure-decay](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0635) [data. Energy Fuels 2005;19\(5\):2041](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0635)–9.
- [128] [Renner TA. Measurement and correlation of di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0640)ffusion coefficients for CO₂ and [rich-gas applications. SPE Reserv Eng 1988;3\(2\):517](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0640)–23.
- [129] [Fu BCH, Phillips CR. New technique for determination of di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0645)ffusivities of volatile [hydrocarbons in semi-solid bitumen. Fuel 1979;58\(8\):557](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0645)–60.
- [130] Das SK, Butler RM. Diffusion coeffi[cients of propane and butane in peace river](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0650) [bitumen. Can J Chem Eng 1996;74\(6\):985](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0650)–92.
- [131] [Wen Y, Bryan J, Kantzas A. Estimation of di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0655)ffusion coefficients in bitumen solvent mixtures as derived from low fi[eld NMR spectra. J Can Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0655) [2005;44\(4\):29](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0655)–34.
- [132] Song L, Kantzas A, Bryan J. Experimental measurement of diffusion coefficient of CO2 in heavy oil using X-ray computed-assisted tomography under reservoir conditions. In: Canadian Unconventional Resources & International Petroleum Conference held in Calgary; 2010.
- [133] Fadaei H, Scarff B, Sinton D. Rapid microfluidics-based measurement of $CO₂$ dif[fusivity in bitumen. Energy Fuels 2011;25\(10\):4829](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0665)–35.
- [134] Schmidt T. Mass transfer by diffusion. In: AOSTRA technical handbook on oil sands, bitumens and heavy oils: Alberta oil sands technology and research authority, Edmonton, Alberta, Canada; 1989. pp. 33–100.
- [135] [Etminan SR, Maini BB, Chen Z, Hassanzadeh H. Constant-pressure technique for](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0675) gas diff[usivity and solubility measurements in heavy oil and bitumen. Energy](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0675) [Fuels 2010;24\(1\):533](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0675)–49.
- [136] Henni A, Shirif E. Solubility and diffusion coefficients of gases in heavy oil and its fractions; 2010.
- [137] [Zhang Y, Hyndman C, Maini B. Measurement of gas di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0685)ffusivity in heavy oils. J Pet [Sci Eng 2000;25:37](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0685)–47.
- [138] Nguyen T, Farouq Ali S. Eff[ect of nitrogen on the solubility and di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0690)ffusivity of [carbon dioxide into oil and oil recovery by the immiscible WAG process. J Can Pet](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0690) [Technol 1998;37\(2\):24](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0690)–31.
- [139] [Behzadfar E, Hatzikiriakos SG. Di](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0695)ffusivity of CO₂ in bitumen: pressure – decay [measurements coupled with rheometry. Energy Fuels 2014;28\(2\):1304](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0695)–11.
- [140] Unatrakarn D, Asghari K, Condor J, Experimental studies of $CO₂$ and CH₄ diffusion coeffi[cient in bulk oil and porous media. Energy Procedia 2011;4:2170](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0700)–7.
- [141] Sankur V, Emanuel AS. A laboratory study of heavy oil recovery with $CO₂$ injection. In: SPE California Regional Meeting, 23-25 March, Ventura, California; 1983.
- [142] Shi R, Kantzas A. Enhanced heavy oil recovery on depleted long core system by CH4 and CO2. In: International Thermal Operations and Heavy Oil Symposium; 2008. p. 20–23.
- [143] Alshmakhy A, Maini B. Eff[ect of foaminess on the performance of solution gas](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0715) [drive in heavy oil reservoirs. J Can Pet Technol 2009;48\(3\):36](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0715)–41.
- [144] [Ekhlasjoo I, Vosoughi M, Shadizadeh SR, Kharrat R, Ghazanfari MH. An experi](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0720)mental and simulation study of heavy oil recovery by the liquid $CO₂$ huff and puff [method. Energy Sources Part A Recover Util Environ E](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0720)ff 2014;36(23):2587–94.
- [145] Sahin S, Kalfa U, Celebioglu D, Duygu E, Lahna H. A quarter century of progress in the application of $CO₂$ immiscible EOR project in Bati Raman heavy oil field in Turkey. In: SPE Heavy Oil Conf. Canada, 12–14 June, Calgary, Alberta, Canada; 2012.
- [146] Olenick S, Schroeder FA, Haines HK, Monger-Mcclure TG. Cyclic CO₂ injection for heavy-oil recovery in halfmoon field: laboratory evaluation and pilot performance. In: SPE Annual Technical Conference and Exhibition, 4-7 October, Washington, D. C.; 1992.
- [147] [Issever K, Pamlr AN, Tlrek A. Performance of a heavy-oil](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0735) field under CO₂ Injection, [Bati Raman, Turkey. SPE Reserv Eng 1993;8\(4\):256](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0735)–60.
- [148] Luo R, Cheng L, Peng J. Feasibility study of $CO₂$ injection for heavy oil reservoir after cyclic steam stimulation: liaohe oil field test. In: Proc. SPEPS-CIM/CHOA Int. Therm. Oper. Heavy Oil Symp., no. 2; 2005.
- [149] Ishii H, Sarma HK, Ono K, Issever K. A successful immiscible CO₂ field pilot in a

carbonate heavy oil reservoir in the Ikiztepe field, Turkey. In: 9th European Symposium on Improved Oil Recovery, The Hague, The Netherlands, 20–22 October; 1997.

- [150] Haskin HK, Alston RB. An evaluation of $CO₂$ huff 'n' puff tests in Texas. J Pet [Technol 1989;41\(2\):177](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0750)–84.
- [151] [Monger TG, Coma JM. A laboratory and](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0755) field evaluation of the CO₂ huff 'n' puff [process for light-oil recovery. SPE Reserv Eng 1988;3\(4\):1168](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0755)–76.
- [152] Thomas GA, MongerMcClure TG. Feasibility of cyclic CO₂ injection for light-oil [recovery. SPE Reserv Eng 1991;6\(2\):179](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0760)–84.
- [153] Brock W, Bryan L. Summary results of $CO₂$ EOR field tests, 1972–1987. In: Low Permeability Reservoirs Symposium, 6-8 March, Denver, Colorado; 1989.
- [154] Palmer FS, Landry RW, Bou-Mikael S. Design and implementation of immiscible carbon dioxide displacement projects (CO₂ huff-puff) in south Louisiana. In: SPE Annual Technical Conference and Exhibition; 1986.
- [155] [Monger TG, Coma JM. A laboratory and](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0775) field evaluation of the $CO₂$ huff 'n' puff [process for light-oil recovery. SPE Reserv Eng 1988;3\(4\):1168](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0775)–76.
- [156] [Patton JT, Sigmund P, Evans B, Ghose S, Weinbrandt D. Carbon dioxide well sti](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0780)mulation: part 2 – design of Aminoil'[s north Bolsa strip project. J Pet Technol](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0780) [1982;34\(8\):1805](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0780)–10.
- [157] Spivak A, Karaoguz D, Issever K, Nolen J. Simulation of immiscible CO₂ injection in a fractured carbonate reservoir, bati raman field, Turkey. In: SPE California Regional Meeting; 1989.
- [158] [Patton JT, Coats KH, Spence K. Carbon dioxide well stimulation: Part 1](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0790) – a [parametric study. J Pet Technol 1982;34\(8\):1798](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0790)–804.
- [159] Colmenares C, Méndez J, E&p P, A new approach on analytical solution for heavy oil recovery by Huff & Puff CO2 injection. Case Study Bachaquero-01 Reservoir", in SPE Latin American and Caribbean Petroleum Engineering Conference; 2015.
- [160] [Marx J, Langenheim R. Reservoir heating by hot](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0800) fluid injection. Pet Trans AIME [1959;216:312](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0800)–5.
- [161] [Boberg TC, Lantz RB. Calculation of the production rate of a thermally stimulated](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0805) [well ABSTRACT. J Pet Technol 1966;18\(12\).](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0805)
- [162] [Mohammed-Singh LJ, Singhal AK. Lessons from Trinidad](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0810)'s CO₂ immiscible pilot [projects. SPE Reserv Eval Eng Oct. 2005;8\(5\):397](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0810)–403.
- [163] Reid TB, Robinson HJ. Lick creek meakin sand unit immiscible $CO₂$ waterflood [project. J Pet Technol 1981;33\(9\):1723](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0815)–9.
- [164] [Zanganeh P, Ayatollahi S, Alamdari A, Zolghadr A, Dashti H, Kord S. Asphaltene](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0820) deposition during CO₂ [injection and pressure depletion: a visual study. Energy](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0820) [Fuels 2012;26\(2\):1412](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0820)–9.
- [165] [Sheu EY, Mullins OC. Asphaltenes: fundamentals and Applications. New York:](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0825) [Plenum Publishing Co.; 1995.](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0825)
- [166] [Leontaritis KJ, Ali Mansoori G. Asphaltene deposition: a survey of](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0830) field experi[ences and research approaches. J Pet Sci Eng 1988;1\(3\):229](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0830)–39.
- [167] [Hashemi-kiasari H, Hemmati-sarapardeh A, Mighani S, Mohammadi AH. E](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0835)ffect of [operational parameters on SAGD performance in a dip heterogeneous fractured](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0835) [reservoir. Fuel 2014;122:82](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0835)–93.
- [168] [Idem RO, Ibrahim HH. Kinetics of CO2-induced asphaltene precipitation from](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0840) [various Saskatchewan crude oils during CO2 miscible](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0840) flooding. J Pet Sci Eng [2002;35\(3](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0840)–4):233–46.
- [169] Huang E. The effect of oil composition and asphaltene content on $CO₂$ displacement. In: SPE/DOE Enhanced Oil Recovery Symposium; 1992.
- [170] Al-Maamari RSH, Buckley JS. Asphaltene precipitation and alteration of wetting: can wettability change during oil production? In: SPE/DOE Improved Oil Recovery Symposium; 2000.
- [171] Yen A, Yin YR, Asomaning S. Evaluating asphaltene inhibitors: laboratory tests and field studies. In: SPE International Symposium on Oilfield Chemistry; 2001.
- [172] [Alomair OA, Almusallam AS. Heavy crude oil viscosity reduction and the impact of](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0860) [asphaltene precipitation. Energy Fuels 2013;27\(12\):7267](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0860)–76.
- [173] Mullins OC, Sheu EY, Hammami A, Marshall AG. Asphaltenes, heavy oils, and petroleomics; 2007.
- [174] [Machado ALC, Lucas EF, González G. Poly\(ethylene-co-vinyl acetate\) \(EVA\) as](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0870) [wax inhibitor of a Brazilian crude oil: oil viscosity, pour point and phase behavior](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0870) [of organic solutions. J Pet Sci Eng 2001;32\(2](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0870)–4):159–65.
- [175] Babadagli T, Sahin S, Kalfa U, Celebioglu D, Karabakal U, Topguder NN. Development of heavy oil fractured carbonate Bati Raman field: evaluation of steam injection potential and improving ongoing CO₂ injection; 2008. p. 1-22.
- [176] [Parker ME, Meyer JP, Meadows SR. Carbon dioxide enhanced oil recovery injec](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0880)[tion operations technologies \(poster presentation\). Energy Procedia](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0880) $2009:1:3141-R$
- [177] Moffitt P, Zornes D. Postmortem analysis: lick creek Meakin Sand Unit Immiscible CO2 waterflood project. In: SPE Annual Technical Conference and Exhibition; 1992.
- [178] [Yuan Q, Azaiez J. Miscible displacements in porous media with time-dependent](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0890) [injection velocities. Transp Porous Med 2014;104:57](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0890)–76.
- [179] [Yuan Q, Azaiez J. Cyclic time-dependent reactive](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0895) flow displacements in porous [media. Chem Eng Sci 2014;109:136](http://refhub.elsevier.com/S0016-2361(17)31495-3/h0895)–46.